

## **Biomolecular Thermodynamics and Kinetics**

BME 4621 Section 18054

**Class Periods:** MWF, 1:55-2:45, 7<sup>th</sup> period

**Location:** LAR 0310

**Academic Term:** FALL 2025

### ***Instructor:***

Gregory A. Hudalla

[ghudalla@bme.ufl.edu](mailto:ghudalla@bme.ufl.edu)

352-273-9326

Office Hours: Monday & Wednesday 12:45-1:45 PM BMS J296

### ***Teaching Assistant/Peer Mentor/Supervised Teaching Student:***

Please contact through the Canvas website

- Joseph Tsenum, [josephtsenum@ufl.edu](mailto:josephtsenum@ufl.edu)

### ***Course Description***

Principles of thermodynamics and kinetics from a biomolecular perspective. The mathematics, analysis, and applications of classical thermodynamics, statistical thermodynamics, and reaction kinetics will be introduced in the context of molecular interactions, binding equilibria, metabolism, and biomolecular transport common to living systems.

### ***Course Pre-Requisites / Co-Requisites***

BME 3060, BME 4311

### ***Course Objectives***

- Develop basic knowledge of classical thermodynamics, equilibrium, and reaction kinetics.
- Develop an understanding of the application of statistical thermodynamics to biomolecule behavior and interactions.
- Develop a competence in the fundamental analytical and computational tools used to describe energy transformation within living systems.

### ***Materials and Supply Fees***

None

### ***Relation to Program Outcomes (ABET):***

Outcome	Coverage*
1. An ability to identify, formulate, and solve complex engineering problems by applying principles of engineering, science, and mathematics	High, emphasized
2. An ability to apply engineering design to produce solutions that meet specified needs with consideration of public health, safety, and welfare, as well as global, cultural, social, environmental, and economic factors	
3. An ability to communicate effectively with a range of audiences	
4. An ability to recognize ethical and professional responsibilities in engineering situations and make informed judgments, which must consider the impact of engineering solutions in global, economic, environmental, and societal contexts	

5. An ability to function effectively on a team whose members together provide leadership, create a collaborative environment, establish goals, plan tasks, and meet objectives	
6. An ability to develop and conduct appropriate experimentation, analyze and interpret data, and use engineering judgment to draw conclusions	
7. An ability to acquire and apply new knowledge as needed, using appropriate learning strategies	

### ***Required Textbooks and Software***

Title: Biomolecular Thermodynamics: From Theory to Application

Author: Barrick, Douglas

Publication date and Edition: 2017, 1<sup>st</sup> edition; CRC Press

ISBN: 978-1-4398-0019-5

### ***Recommended Materials***

Title: *Biological Thermodynamics*

Author: Haynie, Donald T.

Publication date and edition: 2008, 2<sup>nd</sup> edition; Cambridge

ISBN number: 978-0-5217-1134-0

Title: *Molecular Driving Forces: Statistical Thermodynamics in Chemistry and Biology.*

Author: Dill, Ken A., and Bromberg, S.

Publication date: 2002, Routledge

ISBN: 978-0-8153-2051-7.

Title: *Biochemical Engineering Fundamentals*

Author: Bailey, James E., and Ollis, David F.

Publication date: 1986, 2<sup>nd</sup> edition, McGraw-Hill

ISBN: 978-0-0706-6601-6

### ***Required Computer***

Recommended Computer Specifications: <https://it.ufl.edu/get-help/student-computer-recommendations/>

HWCOE Computer Requirements: <https://www.eng.ufl.edu/students/advising/fall-semester-checklist/computer-requirements/>

## Course Schedule

Week	Topic
<u>PART 1</u>	
1	Overview of energy transformation in living systems
2	First law of thermodynamics – Barrick 3
3	Second law of thermodynamics – Barrick 4
4	Gibbs Free Energy – Theory I – Barrick 5
5	Gibbs Free Energy – Theory II
6	Gibbs Free Energy – Applications I
7	Gibbs Free Energy – Applications II
<u>PART 2</u>	
8	Statistical Thermodynamics - Theory – Barrick 8, 9
9	Statistical Thermodynamics - Application
10	Binding Equilibria – Barrick 13, 14
11	Cooperativity/Allostery
12	Reaction kinetics - Theory
13	Reaction kinetics – Modeling I
14	Reaction kinetics – Modeling II
15	Biology, complexity, and evolution

## Evaluation of Grades

Assignment	Total Points	Percentage of Final Grade
Homework Sets (6)	20 each	40%
Exams (3)	60 each	60%
	300	100%

### Assessment dates:

**Problem sets** (due Wednesdays by midnight via Canvas): Sept 3, Sept 17, Oct 8, Oct 29, Nov 19, Dec 3

**Exams** (Fridays, in class, via Canvas): Sept 26, Nov 7, Dec 12

## Grading Policy

Percent	Grade	Grade Points
94 - 100	A	4.00
90.0 - 93.99	A-	3.67
87 - 89.99	B+	3.33
83 - 86.99	B	3.00
80 - 82.99	B-	2.67
77 - 79.99	C+	2.33
73 - 76.99	C	2.00
70 - 72.99	C-	1.67
67 - 69.99	D+	1.33
63 - 66.99	D	1.00
60 - 62.99	D-	0.67
0 - 59.99	E	0.00

***Academic Policies & Resources***

To support consistent and accessible communication of university-wide student resources, instructors must include this link to academic policies and campus resources: <https://go.ufl.edu/syllabuspolices>. Instructor-specific guidelines for courses must accommodate these policies.

***Commitment to a Positive Learning Environment***

The Herbert Wertheim College of Engineering values varied perspectives and lived experiences within our community and is committed to supporting the University's core values.

If you feel like your performance in class is being impacted, please contact your instructor or any of the following:

- Your academic advisor or Undergraduate Coordinator
- HWC OE Human Resources, 352-392-0904, [student-support-hr@eng.ufl.edu](mailto:student-support-hr@eng.ufl.edu)
- Pam Dickrell, Associate Dean of Student Affairs, 352-392-2177, [pld@ufl.edu](mailto:pld@ufl.edu)